

(FILE 'HOME' ENTERED AT 13:22:00 ON 08 OCT 1998)

FILE 'INSPEC' ENTERED AT 13:22:32 ON 08 OCT 1998

L1 9002 CADMIUM TELLURIDE OR CDTE  
L2 11321 CDS OR CADMIUM SULPHIDE  
L3 67021 ARGON OR AR  
L4 12266 L3(2A) (ION# OR ATOM#)  
L5 59 L2 AND L3 AND L4  
L6 18816 OHM###  
L7 1 L5 AND L6  
L8 4 L1 AND L2 AND L4

FILE 'CA' ENTERED AT 13:29:25 ON 08 OCT 1998

L9 6 L8

FILE 'WPIDS' ENTERED AT 13:43:19 ON 08 OCT 1998

L10 1 L8

FILE 'INSPEC' ENTERED AT 13:44:10 ON 08 OCT 1998

FILE 'WPIDS' ENTERED AT 13:44:13 ON 08 OCT 1998

L11 18126 P TYPE  
L12 64 P (2A)L1  
L13 0 L12 AND L4

FILE 'INSPEC' ENTERED AT 13:47:41 ON 08 OCT 1998

L14 4 L13

FILE 'CA' ENTERED AT 13:50:19 ON 08 OCT 1998

L15 4 L13  
L16 0 L14 NOT L15

L8 ANSWER 1 OF 4 INSPEC COPYRIGHT 1998 IEE

AB. . . host materials. Two of them are attributed to glass defects around semiconductor nanocrystals and the other to S vacancies in **Cds** nanocrystals. The increase in electron spin resonance absorption with **Ar-ion** laser irradiation reveals that carriers photoexcited in nanocrystals are trapped not only on nanocrystal defects but also glass defects.

ST semiconductor-doped glasses; ESR; phosphate glasses; precipitation; nanocrystals; S vacancies; **Ar-ion laser irradiation**; 77 K; **P205-Cds**; **P205-CdSe**; **P205-CdTe**

L8 ANSWER 2 OF 4 INSPEC COPYRIGHT 1998 IEE

TI Origin and identification of impurities in electrodeposited **cadmium telluride** films.

AB Impurities in electrodeposited thin films of **CdTe** are identified from secondary ion mass spectroscopy (SIMS) data obtained with a Perkin-Elmer PHI Model 2500, a Cameca IMS 3F and a Cameca IMS 4F operated with the primary ions **Ar+**, **O-** and **Cs+** respectively. The sources of the impurities are sought using the techniques of SIMS, inductively coupled plasma-atomic emission. . . atomic absorption spectrophotometry to analyse the components of the plating bath solution (cadmium sulphate, Milli-Q water, sulphuric acid, tellurium anode, glass/ITO/**Cds**-cathode, reference electrodes) and the solution containers. Six different supplies of **3CdSO4.8H2O** are investigated. Each had at least twelve impurities totalling over 61  $\mu$ g/g. Each impurity in the **CdTe** films is present in the deposition chemicals or leached from the materials in contact with these chemicals. It is concluded that **CdTe** film purity may be enhanced by judicious choice of deposition materials and procedure.

ST. . . inductively coupled plasma-atomic emission spectroscopy; atomic absorption spectrophotometry; plating bath solution; Milli-Q water; reference electrodes; solution containers; **3CdSO4.8H2O**; deposition chemicals; **electrodeposited CdTe thin films**; **CdSO4H2O**

L8 ANSWER 3 OF 4 INSPEC COPYRIGHT 1998 IEE

AB. . . scanning electron microscope after bombardment at normal incidence at room temperature with doses between  $0.5 \times 10^{18}$  and  $3 \times 10^{18}$  of 40 keV **argon ions/cm2**. The usual topography of the sputtered surface and the perturbing effects of dirt and contamination are reported, and observations of. . . steep slopes. The behaviour of Si, Ge, GaAs and InP was consistent with an amorphous surface. In the case of **CdTe**, **Cds** and GaP it appeared that the crystal structure was retained resulting in a topography of the sputtered surface that is. . .

ST sputtered semiconductors; cone formation; GaAs; flux enhancement; Si; Ge; GaAs; InP; **CdTe**; **Cds**; GaP; **Ar ions**

L8 ANSWER 4 OF 4 INSPEC COPYRIGHT 1998 IEE

AB The structure of (1010), (1120) planes and the (0001) Cd, (0001) S polar surfaces of AIIBVI-compounds: **Cds**, **CdSe** and (110) **CdTe** planes grown from the vapor phase were studied by means of optic and scanning electron microscope. The etching of crystals by 1.5 keV **Ar+ ion** bombardment was carried out. The analysis by microanalyzer 'Cameca' showed that the phase

ST

composition of the crystals before and after. .  
group II VI compounds; semiconductors; surface structure;  
sputtering; ions scattering; **CdS**; CdSe; **CdTe**;  
etching; polar surface identification; phase composition; ion  
bombardment; low angle boundaries; dislocation density; glide bands

(FILE 'USPAT' ENTERED AT 13:56:27 ON 08 OCT 1998)

L1 2835 CADMIUM TELLURIDE OR CDTE  
L2 9404 CADMIUM SULPHIDE OR CDS  
L3 187 P(3A)L1  
L4 111894 ARGON OR AR  
L5 8087 L4(3A)(ION# OR ATOM#)  
L6 7 L3 AND L5 AND L2  
L7 28 DHIS  
L8 13 L3 AND L5  
L9 0 L6 NOT L8  
L10 6 L8 NOT L6  
L11 160 438/525,528/CCLS  
L12 0 L5 AND L1 AND L11  
L13 1 L1 AND L11  
L14 22 L5 AND L11  
L15 33 438/518/CCLS  
L16 1 L5 AND L1 AND L15  
L17 0 L1 AND L4 AND L11  
L18 66 438/718/CCLS  
L19 3 L1 AND L5 AND L18  
L20 39 438/603/CCLS  
L21 1 L1 AND L5 AND L20  
L22 0 CDS/CDTE  
L23 3 L5 AND L20  
L24 1 L23 AND L1

=> d 119 1-3

1. 5,318,666, Jun. 7, 1994, Method for via formation and type conversion in group II and group VI materials; Jerome L. Elkind, et al., **438/718**, 513, 916, 971 [IMAGE AVAILABLE]

2. 5,017,511, May 21, 1991, Method for dry etching vias in integrated circuit layers; Jerome L. Elkind, et al., 438/704; 148/DIG.51; 216/67, 75; 427/534; **438/718** [IMAGE AVAILABLE]

3. 4,411,732, Oct. 25, 1983, Method of manufacturing a detector device; John T. M. Wotherspoon, 204/192.34; 257/442, 461; 427/528; 438/518, 571, **718**, 971 [IMAGE AVAILABLE]

=> d 123 1-3

1. 5,213,998, May 25, 1993, Method for making an ohmic contact for p-type group II-VI compound semiconductors; Jun Qiu, et al., 438/46; 117/108; 148/DIG.64, DIG.95; **438/603** [IMAGE AVAILABLE]

2. 4,439,912, Apr. 3, 1984, Infrared detector and method of making same; John H. Pollard, et al., 438/72; 204/192.17, 192.34; 250/370.13; 257/442, 448; 438/98, **603** [IMAGE AVAILABLE]

3. 3,983,264, Sep. 28, 1976, Metal-semiconductor ohmic contacts and methods of fabrication; Walter H. Schroen, et al., 257/734; 204/192.17, 192.3, 192.34; 257/473; 427/299, 309, 527, 578; 438/597, 602, **603**, 604 [IMAGE AVAILABLE]